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Introduction

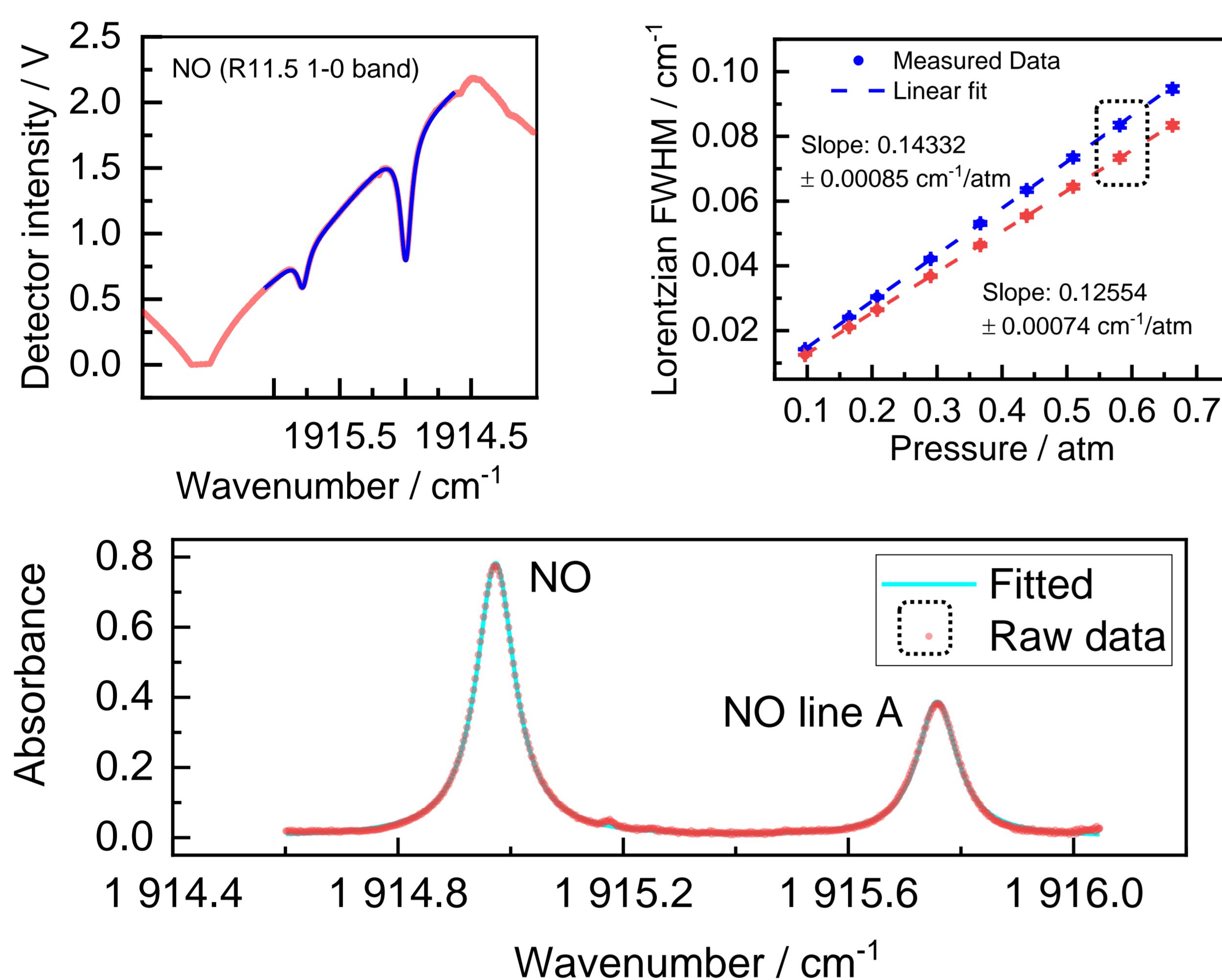
- To improve **hydrogen**-related combustion processes, thorough knowledge of their underlying **reaction kinetics** is essential
- NO_x**-emissions (only harmful byproducts of H₂ combustion), need to be measured more accurately
- Time resolved **temperature**, **pressure** and **concentration** measurements are needed to validate kinetic models

Solution:

Tunable Diode Laser Absorption Spectroscopy (TDLAS)
→ requires **high quality traceable data** of pressure broadening coefficients for foreign molecules (O₂, H₂, CO₂, Ar, He, N₂)

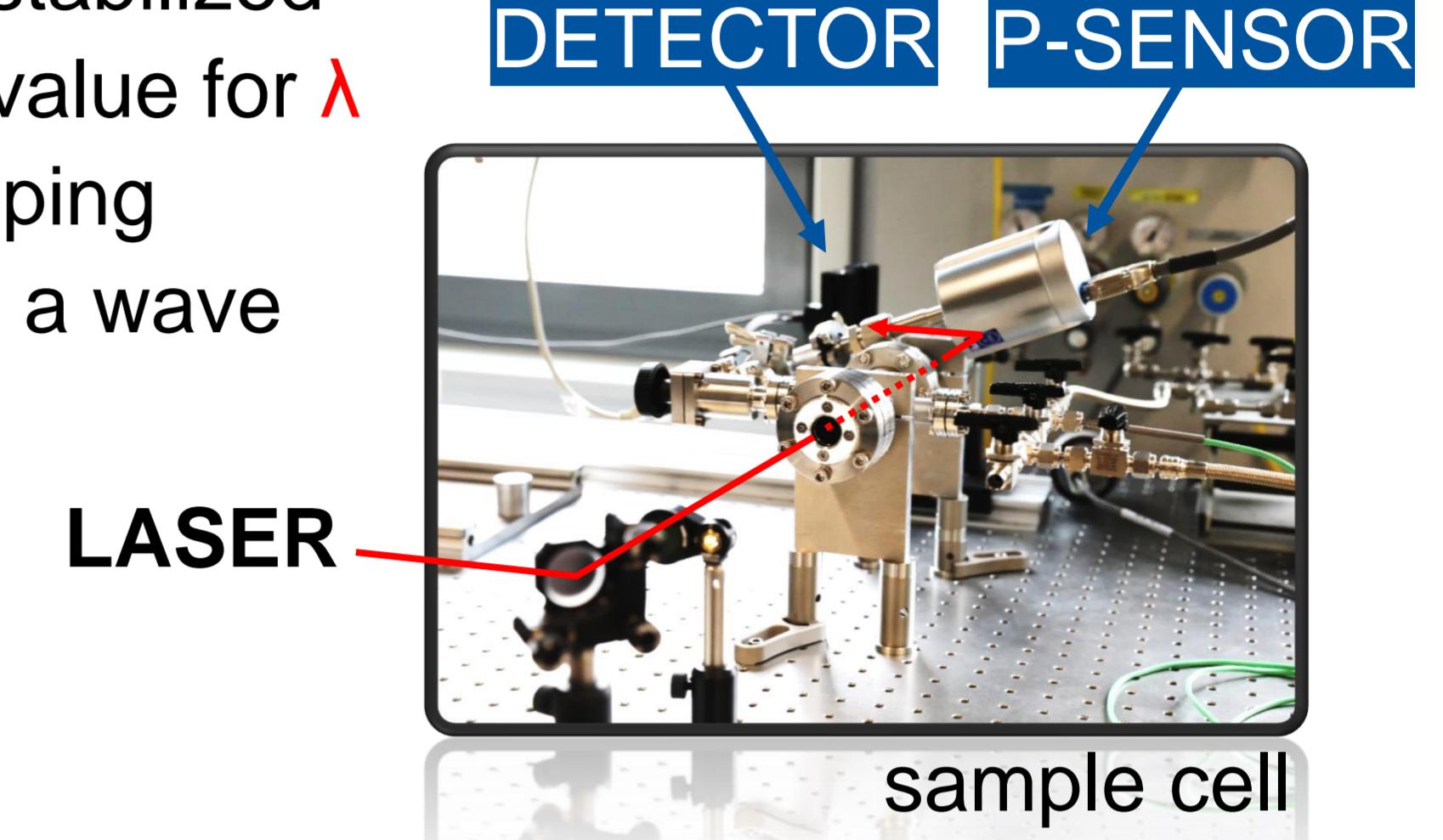
NO spectrum in H₂

- center wavelength of Laser: 5220 nm
- 2 peaks** measured: NO [1914.9 cm⁻¹], NO (A) [1915.7 cm⁻¹]



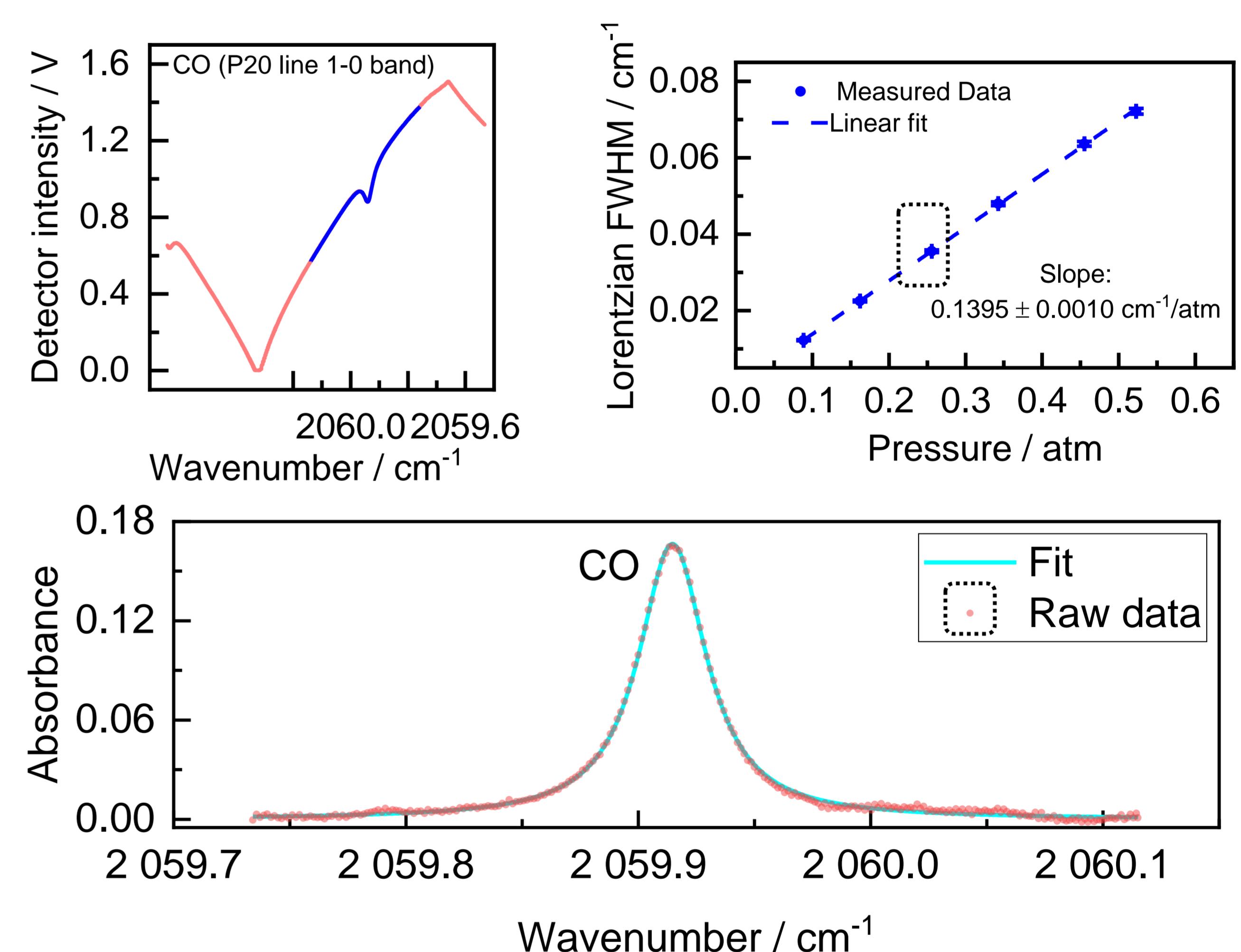
Setup

- single pass cell, absorption path **300 mm**, pressure **0 - 1 bar**
- Mid-IR range **Interband Cascade Lasers** (custom center WN)
- stable** single mode emission and **mode-hop-free**
- Laser diode temperature stabilized by TEC and set to target value for λ
- wavelength tuning by ramping the laser current I_{LD} using a wave function generator



CO spectrum in H₂

- center wavelength of Laser: 4855 nm
- 1 line** measured: CO [2059.9 cm⁻¹]



Pressure Broadening Coefficients γ

$$\Delta\nu = 2 c_{gas} p_{gas} \gamma_{self} + 2 \sum_{i=1}^n c_i p_i \gamma_i$$

Self Broadening + Foreign Broadening

- Consistency**: all measurements acquired within the same setup
- Traceability**: laser tuning (etalon), pressure, temperature
- UNIT**: 10⁻³ cm⁻¹/atm

Absorption Line	Ar	He	H ₂	O ₂	N ₂	CO ₂	Air
NO	42.68	44.06	71.66	60.39	57.60	-	-
NO (A)	37.38	37.82	62.77	58.31	50.53	-	-
CO	38.92	48.53	69.75	44.51	53.25	57.13	52.37

Conclusions

- new data provides a **base for future investigations** using laser diagnostics such as TDLAS
- Knowledge of broadening coefficients with **low uncertainty** improves simulation of spectra, that are needed to obtain absolute concentration measurements

References

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